

## A Convenient One-Pot Preparative Method for 4,5-Diarylisoxazoles Involving Amine Exchange Reactions

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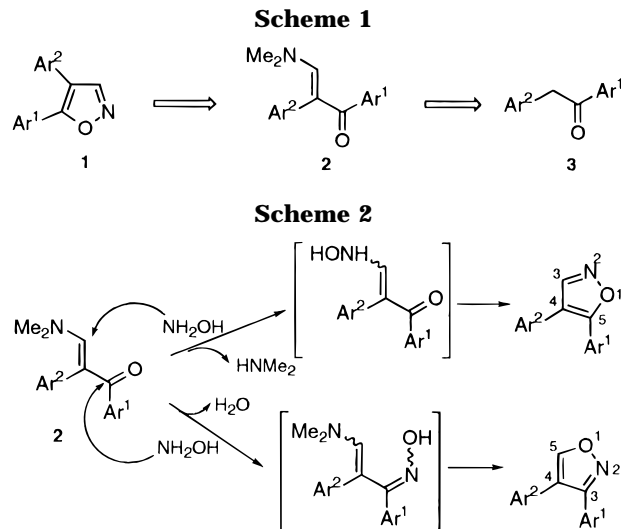
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4,5-Diarylisoxazoles **1** are efficiently prepared by submitting enamino ketones **2**, readily obtained from deoxybenzoins **3**, to oximation conditions. This conversion implies an uncommon amine group exchange reaction. Preparation of 4-isoxazolines **5** and ring-opening transformations to  $\beta$ -keto nitriles **4** and 1,3-amino alcohol derivatives **7** are also described.

In the context of our ongoing interest in the synthetic potential of enamino ketones **2**,<sup>1</sup> we set out to study the synthesis and reactivity of 4,5-diarylisoxazoles **1**<sup>2</sup> using compounds **2** as key intermediates (Scheme 1).

Enamino ketones,<sup>3</sup>  $\beta$ -diketones,<sup>4</sup> and  $\beta$ -chlorovinyl ketones<sup>5</sup> have been used *inter alia* as starting materials for the preparation of isoxazoles.<sup>6</sup> In every case, the cyclization reaction can be typified as C–C–C + N–O ring closure process. However, only the use of enamino ketone provides a single isomer,<sup>7</sup> which is the result of an unusual amine exchange reaction.<sup>1b,8</sup> Nevertheless, in all cases the cyclization reaction may proceed by two possible mechanisms,<sup>9</sup> which differ in their sequential nucleophilic attack/amine exchange reactions (Scheme 2). As such, the identification of the reaction products provides unequivocal proof<sup>4a,9</sup> of the mechanism involved in the transformation.

The isoxazole ring has wide synthetic applications.<sup>10</sup> Particularly, a 4,5-diaryl substituted heterocycle adequately functionalized can be used as a substrate for Ullmann or Suzuki diaryl coupling reactions,<sup>11–13</sup> leading to tetracyclic planar systems of great interest. Therefore, and taking into account the few examples of this type of



heterocycle in the literature,<sup>14</sup> it appeared worthwhile to find a simple high-yielding method for the synthesis of 4,5-diaryl-substituted isoxazoles.

### Results and Discussion

Enamino ketones **2a–g**, readily obtained from the corresponding deoxybenzoins **3**,<sup>1</sup> were submitted to reaction with hydroxylamine hydrochloride under standard oximation conditions,<sup>15</sup> yielding 4,5-diarylisoxazoles **1a–g** in a “one-pot” reaction, as shown in Table 1 and Scheme 3.

The results allowed us to propose a mechanism for the formation of the target isoxazoles in which the first step is an amine exchange reaction (Scheme 4). Subsequent nucleophilic attack of the so-obtained hydroxylamine derivative to the carbonyl group, followed by elimination, yields the target heterocycle.

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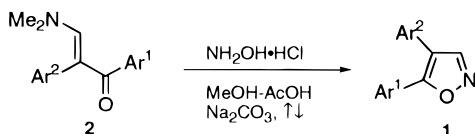
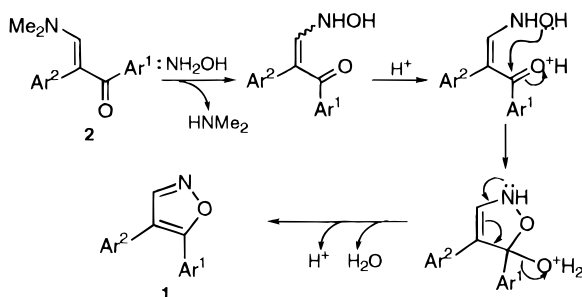
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**Table 1.** 4,5-Diarylisoaxazoles **1** Prepared

isoxazoles	Ar <sup>1</sup>	Ar <sup>2</sup>	yield <sup>a</sup> (%)	mp <sup>b</sup> (°C)
<b>1a</b>	Ph	Ph	90	68–70 <sup>14</sup>
<b>1b</b>	3,4-(MeO) <sub>2</sub> Ph	3,4-(MeO) <sub>2</sub> Ph	99	126–128
<b>1c</b>	3,4-(MeO) <sub>2</sub> Ph	Ph	91	62–63
<b>1d</b>	2,3,4-(MeO) <sub>3</sub> Ph	3,4-(MeO) <sub>2</sub> Ph	96	52–54
<b>1e</b>	2,3,4-(MeO) <sub>3</sub> Ph	Ph	97	69–71
<b>1f</b>	3,4-(MeO) <sub>2</sub> Ph	3,4,5-(MeO) <sub>3</sub> Ph	95	82–84
<b>1g</b>	2,3,4-(MeO) <sub>3</sub> Ph	3,4,5-(MeO) <sub>3</sub> Ph	88	oil

<sup>a</sup> Yield of pure crystallized compound. <sup>b</sup> Crystallized from methanol.

**Scheme 3****Scheme 4****Table 2.**  $\beta$ -Ketonitriles **4** Prepared

$\beta$ -ketonitriles	Ar <sup>1</sup>	Ar <sup>2</sup>	yield <sup>a</sup> (%)	mp <sup>b</sup> (°C)
<b>4a</b>	Ph	Ph	70	81–82
<b>4b</b>	3,4-(MeO) <sub>2</sub> Ph	3,4-(MeO) <sub>2</sub> Ph	91	103–105
<b>4c</b>	3,4-(MeO) <sub>2</sub> Ph	Ph	81	113–114
<b>4d</b>	2,3,4-(MeO) <sub>3</sub> Ph	3,4-(MeO) <sub>2</sub> Ph	88	114–115

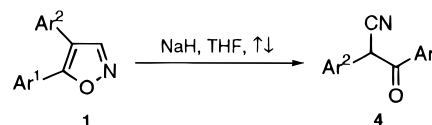
<sup>a</sup> Yield of pure crystallized compound. <sup>b</sup> Crystallized from hexane–diethyl ether (1:1).

Routinely, basic treatment of C-3-unsubstituted isoxazoles induces ring opening to give  $\beta$ -keto nitriles,<sup>6,16</sup> useful difunctionalized compounds.<sup>17</sup> In our case, treatment of heterocycles **1** with EtONa/EtOH<sup>17b</sup> under various reaction conditions provided the expected  $\beta$ -keto nitriles **4** in 30–36% yield. Low yields (20–28%) also resulted as well by using LDA as a base, although it has been reported that this reagent has been successfully applied to 5-substituted isoxazoles.<sup>16b</sup> Furthermore, though LiAlH<sub>4</sub> has been reported to selectively reduce isoxazoles,<sup>18c</sup> in our hands  $\beta$ -keto nitriles **4** were obtained via a ring-opening process, only with low yields (19–25%). Finally,  $\beta$ -keto nitriles **4** were prepared in high yield using NaH (Table 2, Scheme 5).

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**Scheme 5**

In the light of these results, we propose that the observed ring-opening reaction not only represents a practical synthesis of the mentioned ketonitriles, but also a valuable method for checking the position of the substituents in the parent heterocycle,<sup>3,9a</sup> thus establishing unambiguously the mechanism previously proposed for the construction of isoxazoles **1**. In fact, only the 4,5-substituted isoxazole, and not the 3,4-disubstituted isoxazole, would be able to react with base to directly produce a cyanoketone.

Similar to isoxazoles,<sup>18</sup> isoxazolines also have pharmacological,<sup>19</sup> industrial,<sup>20</sup> and synthetic applications.<sup>21</sup> Among these compounds, 4-isoxazolines **5**<sup>2,22</sup> are quite rare, and as far as we know, no general method for their preparation has been reported to date. In order to perform the selective reduction of isoxazoles **1** to 4-isoxazolines **5**, we first carried out *N*-methylation<sup>17a,23</sup> reactions with trimethyloxonium tetrafluoroborate (Meerwein salt).<sup>24</sup> The so-obtained *N*-methylisoxazolinium salts **6** were then submitted to reduction with both LiAlH<sub>4</sub> and NaBH<sub>4</sub>,<sup>22</sup> affording quite different results. In fact, while low yields of 4-isoxazolines **5** were obtained with the former, the use of NaBH<sub>4</sub> considerably improved the outcome (Table 3, Scheme 6).

On the other hand, although 2-isoxazolines are known to be good precursors<sup>22,25,26</sup> of 1,3-amino alcohols, versatile building blocks for natural product synthesis<sup>27</sup> as well as useful chiral inductors,<sup>28</sup> no examples of reductive cleavage of 4-isoxazolines of type **5** to 1,2-diaryl-substituted 1,3-amino alcohols **7** are known. Toward this goal, we attempted the tandem reduction of the stilbenic C–C double bond/cleavage of the N–O bond of isoxazolines **5**,

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Table 3. Isoxazolinium Salts 6 and 4-Isoxazolines 5 Prepared

	Ar <sup>1</sup>	Ar <sup>2</sup>	6		5	
			yield <sup>a</sup> (%)	mp <sup>b</sup> (°C)	yield <sup>a</sup> (%)	mp (°C)
a	Ph	Ph	85	139–141	84	oil
b	3,4-(MeO) <sub>2</sub> Ph	3,4-(MeO) <sub>2</sub> Ph	94	161–163	64	109–111 <sup>c</sup>
c	3,4-(MeO) <sub>2</sub> Ph	Ph	96	158–160	90	oil
d	2,3,4-(MeO) <sub>3</sub> Ph	3,4-(MeO) <sub>2</sub> Ph	99	152–153	93	oil

<sup>a</sup> Yield of pure compound. <sup>b</sup> Crystallized from ethyl acetate. <sup>c</sup> Crystallized from methanol.

Scheme 6

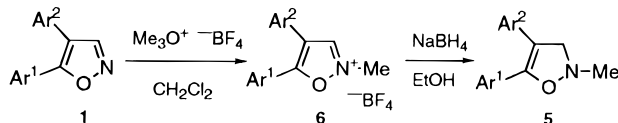
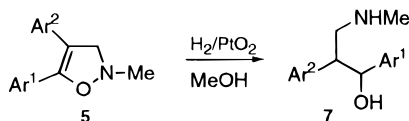


Table 4. 1,3-Amino Alcohols 7 Prepared

1,3-amino alcohols	Ar <sup>1</sup>	Ar <sup>2</sup>	yield <sup>a</sup> (%)	mp (°C)
7a	Ph	Ph	47	oil
7b	3,4-(MeO) <sub>2</sub> Ph	3,4-(MeO) <sub>2</sub> Ph	55	117–119 <sup>b</sup>

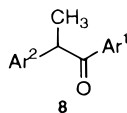
<sup>a</sup> Yield of pure compound. <sup>b</sup> Crystallized from methanol.

Scheme 7



but neither BBN<sup>29</sup> nor DIBAL-H,<sup>26b,30</sup> under a wide range of conditions, induced reaction.

Nevertheless, while Pd–C catalytic hydrogenation only produced 1,2-diarylpropanones **8**, the use of the Adams



catalyst furnished the expected 1,3-amino alcohols **7** though in moderate yields (47% and 55%) (Table 4, Scheme 7).

To summarize, we have accomplished an efficient conversion of deoxybenzoin to 4,5-diarylisoxazoles and we have studied their behaviour under ring opening reaction conditions in basic medium, thus isolating the corresponding  $\beta$ -ketonitriles, which represents a proof of the structure of the original heterocycles. We have also prepared a series of 4,5-diaryl-4-isoxazolines by submitting the parent heterocycles to tandem *N*-quaternization/reduction conditions. Apart from that, the so-obtained isoxazolines were regioselectively converted to the corresponding 1,2-diaryl-substituted 1,3-aminoalcohols.

## Experimental Section

**General Procedures.** Solvents were either purified according to methods described by Perrin *et al.*<sup>31</sup> or used as received from the manufacturers, depending on their purity. Thin layer chromatography (TLC) was performed on plates coated to a thickness of 0.2 mm with Merck Kieselgel 60 F<sub>254</sub>

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using UV light (254 nm) and Dragendorff's reagent<sup>32</sup> as developing agents. Flash column chromatography<sup>33</sup> was performed on Merck Kieselgel 60 (230–400 mesh ASTM); air-pressure chromatography was carried out on Kieselgel 60 (70–230 mesh ASTM) or on neutral alumina 90 (activity I, 70–230 mesh ASTM). Evaporation of solvents under reduced pressure was performed in a rotatory evaporator. Melting points are uncorrected. Infrared spectra were recorded as KBr plates, as a neat liquid, or in CHCl<sub>3</sub>, and peaks are reported in cm<sup>-1</sup>. NMR spectra were recorded at 250 MHz for <sup>1</sup>H and 62.83 MHz for <sup>13</sup>C. Chemical shifts ( $\delta$ ) were measured in ppm relative to tetramethylsilane ( $\delta$  0.00) or chloroform ( $\delta$  7.26 for <sup>1</sup>H or 77.00 for <sup>13</sup>C) as internal standards; dimethyl sulfoxide-*d*<sub>6</sub> ( $\delta$  2.49 for <sup>1</sup>H or 39.5 for <sup>13</sup>C) has been used when indicated. Multiplicities are indicated by s (singlet), bs (broad singlet), d (doublet), t (triplet), q (quartet), m (multiplet), or dd (doublet of doublets). Coupling constants, *J*, are reported in hertz. <sup>13</sup>C DEPT experiments were used to assist with the assignment of the signals. Data for mass spectra (EI) are reported in the form *m/z* (intensity relative to base = 100).

Enamino ketones **2e** and **2g** were prepared in accordance with our previous reports,<sup>1</sup> and the following results were obtained.

**3-(*N,N*-Dimethylamino)-2-phenyl-1-(2,3,4-trimethoxyphenyl)propenone (2e)** (77%): pale yellow powder; mp 137–138 °C (diethyl ether); *R<sub>f</sub>* (hexane–EtOAc, 2:8) 0.4;  $\nu_{\max}$  1640 (C=O),  $\delta_{\text{H}}$  2.60 (6H, s, NMe<sub>2</sub>), 3.78 (6H, s, OMe), 3.83 (3H, s, OMe), 6.64 (1H, d, *J* = 8.5, H-5'arom), 6.87 (1H, d, *J* = 8.5, H-6'arom), 7.11 (1H, s, =CHN), 7.1–7.2 (5H, m, Harom);  $\delta_{\text{C}}$  43.1 (NMe), 55.7, 60.6, and 61.6 (OMe), 106.6, 113.1 (C<sub>arom</sub>H), 123.1 (=CCO), 126.0, 127.2, and 131.8 (C<sub>arom</sub>H), 129.5, 136.4 (C<sub>arom</sub>C), 141.5, 150.5, and 153.5 (C<sub>arom</sub>O), 153.9 (=CHN), 192.6 (CO). Anal. Calcd for C<sub>20</sub>H<sub>23</sub>O<sub>4</sub>N: C, 70.36; H, 6.79; N, 4.10. Found: C, 70.45; H, 6.85; N, 4.20.

**3-(*N,N*-Dimethylamino)-1-(2,3,4-trimethoxyphenyl)-2-(3,4,5-trimethoxyphenyl)propenone (2g)** (88%): pale yellow powder; mp 139–141 °C (diethyl ether); *R<sub>f</sub>* (hexane–EtOAc, 2:8) 0.1;  $\nu_{\max}$  1635 (C=O);  $\delta_{\text{H}}$  2.67 (6H, s, NMe<sub>2</sub>), 3.77 (9H, s, OMe), 3.81 (6H, s, OMe), 3.85 (3H, s, OMe), 6.39 (2H, s, H-2''arom and H-6''arom), 6.58 (1H, d, *J* = 8.5, H-5'arom), 6.87 (1H, d, *J* = 8.5, H-6'arom), 7.07 (1H, s, =CHN);  $\delta_{\text{C}}$  43.3 (NMe), 55.9, 60.6, 60.7, and 61.8 (OMe), 106.7, 109.1 (C<sub>arom</sub>H), 113.3 (=CCO), 123.1 (C<sub>arom</sub>H), 129.5, 131.9 (C<sub>arom</sub>C), 136.4, 141.8, 150.7, 152.2, and 153.7 (C<sub>arom</sub>O), 154.1 (=CHN), 192.9 (CO). Anal. Calcd for C<sub>23</sub>H<sub>29</sub>O<sub>7</sub>N: C, 64.02; H, 6.77; N, 3.25. Found: C, 63.96; H, 6.87; N, 3.58.

**4,5-Diphenylisoxazole (1a).** **Typical Procedure.** NH<sub>2</sub>·OH·HCl (210 mg, 3.15 mmol) and Na<sub>2</sub>CO<sub>3</sub> (170 mg, 1.6 mmol) were added to a stirred solution of enamino ketone **2a**<sup>1</sup> (720 mg, 2.86 mmol) in methanol (30 mL) and water (15 mL) at room temperature. The reaction mixture was acidified with glacial acetic acid up to pH 4–5 and refluxed for 2 h. After cooling, the reaction mixture was basified to pH 8 with ammonium hydroxide solution and extracted with dichloromethane (4 × 30 mL). The organic extracts were dried over anhydrous sodium sulfate and evaporated *in vacuo* to afford a yellow oil, which was crystallized in methanol, providing isoxazole **1a** (570 mg, 90%) as a white powder: mp 68–70 °C (methanol); *R<sub>f</sub>* (CH<sub>2</sub>Cl<sub>2</sub>–EtOAc, 9:1) 0.8;  $\nu_{\max}$  1630 (C=N);  $\delta_{\text{H}}$  7.3–7.4 (8H, m, Harom), 7.6–7.7 (2H, m, Harom), 8.34 (1H, s,

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H-3);  $\delta_C$  116.1 (C-4), 127.2 ( $C_{\text{arom}}\text{H}$ ), 127.5 ( $C_{\text{arom}}\text{C}$ ), 127.9, 128.5, 128.7, and 128.9 ( $C_{\text{arom}}\text{H}$ ), 129.9 ( $C_{\text{arom}}\text{C}$ ), 151.7 (C-3), 163.9 (C-5);  $m/z$  221 ( $M^+$ , 23), 193 (22), 165 (60), 143 (11), 116 (15), 105 (55), 77 (32). Anal. Calcd for  $C_{15}H_{11}ON$ : C, 81.42; H, 5.01; N, 6.33. Found: C, 81.39; H, 4.98; N, 6.40.

By use of the same procedure the following isoxazoles were prepared:

**4,5-Bis(3,4-dimethoxyphenyl)isoxazole (1b)** (99%): mp 126–128 °C (methanol);  $R_f$  ( $\text{CH}_2\text{Cl}_2$ –EtOAc, 9:1) 0.8;  $\nu_{\text{max}}$  1630 ( $\text{C}=\text{N}$ );  $\delta_{\text{H}}$  3.76 (3H, s, OMe), 3.79 (3H, s, OMe), 3.89 (3H, s, OMe), 3.91 (3H, s, OMe), 6.8–7.0 (4H, m,  $H_{\text{arom}}$ ), 7.2–7.3 (2H, m,  $H_{\text{arom}}$ ), 8.29 (1H, s, H-3);  $\delta_C$  55.8, 55.9 (OMe), 109.9, 110.9, 111.5, and 111.9 ( $C_{\text{arom}}\text{H}$ ), 114.9 (C-4), 119.3, 121.3 ( $C_{\text{arom}}\text{H}$ ), 122.8 ( $C_{\text{arom}}\text{C}$ ), 148.8, 149.1, and 150.3 ( $C_{\text{arom}}\text{O}$ ), 151.9 (C-3), 163.5 (C-5);  $m/z$  341 ( $M^+$ , 34), 298 (10), 165 (100), 137 (8), 77 (3). Anal. Calcd for  $C_{19}H_{19}O_5N$ : C, 66.85; H, 5.61; N, 4.10. Found: C, 66.70; H, 5.52; N, 4.01.

**5-(3,4-Dimethoxyphenyl)-4-phenylisoxazole (1c)** (91%): mp 62–64 °C (methanol);  $R_f$  ( $\text{CH}_2\text{Cl}_2$ –EtOAc, 9:1) 0.7;  $\nu_{\text{max}}$  1620 ( $\text{C}=\text{N}$ );  $\delta_{\text{H}}$  3.58 (3H, s, OMe), 3.75 (3H, s, OMe), 6.71 (1H, d,  $J = 8.4$ , H-5'arom), 7.02 (1H, d,  $J = 1.9$ , H-2'arom), 7.12 (1H, dd,  $J = 8.4$ , 1.9, H-6'arom), 7.2–7.3 (5H, m,  $H_{\text{arom}}$ ), 8.21 (1H, s, H-3);  $\delta_C$  55.1, 55.3 (OMe), 109.5, 110.7 ( $C_{\text{arom}}\text{H}$ ), 114.6 (C-4), 119.7, 119.8, 127.5 and 128.3 ( $C_{\text{arom}}\text{H}$ ), 128.4, 129.9 ( $C_{\text{arom}}\text{C}$ ), 148.4, 149.9 ( $C_{\text{arom}}\text{O}$ ), 151.3 (C-3), 163.4 (C-5);  $m/z$  281 ( $M^+$ , 100), 238 (10), 195 (4), 166 (8), 165 (55), 116 (5). Anal. Calcd for  $C_{17}H_{15}O_3N$ : C, 72.58; H, 5.37; N, 4.98. Found: C, 71.98; H, 5.33; N, 5.13.

**4-(3,4-Dimethoxyphenyl)-5-(2,3,4-trimethoxyphenyl)isoxazole (1d)** (96%): mp 98–100 °C (methanol);  $R_f$  ( $\text{CH}_2\text{Cl}_2$ –EtOAc, 9:1) 0.6;  $\nu_{\text{max}}$  1620 ( $\text{C}=\text{N}$ );  $\delta_{\text{H}}$  3.51 (6H, s, 2 × OMe), 3.70 (9H, s, OMe), 6.5–6.9 (5H, m,  $H_{\text{arom}}$ ), 8.34 (1H, s, H-3);  $\delta_C$  55.0, 55.2, 55.5, 60.3, and 60.7 (OMe), 106.8, 110.0, 110.8 ( $C_{\text{arom}}\text{H}$ ), 114.2 (C-4), 116.7 ( $C_{\text{arom}}\text{C}$ ), 119.2 ( $C_{\text{arom}}\text{H}$ ), 122.1 ( $C_{\text{arom}}\text{C}$ ), 125.0 ( $C_{\text{arom}}\text{H}$ ), 141.8, 147.9, and 148.4 ( $C_{\text{arom}}\text{O}$ ), 149.7 (C-3), 151.7, 155.1 ( $C_{\text{arom}}\text{O}$ ), 161.4 (C-5);  $m/z$  371 ( $M^+$ , 32), 238 (10), 340 (5), 195 (100), 152 (7), 77 (3). Anal. Calcd for  $C_{20}H_{21}O_6N$ : C, 64.68; H, 5.70; N, 3.77. Found: C, 64.58; H, 5.73; N, 4.02.

**4-Phenyl-5-(2,3,4-trimethoxyphenyl)isoxazole (1e)** (97%): mp 69–71 °C (methanol);  $R_f$  ( $\text{CH}_2\text{Cl}_2$ –EtOAc, 9:1) 0.8;  $\nu_{\text{max}}$  1620 ( $\text{C}=\text{N}$ );  $\delta_{\text{H}}$  3.57 (3H, s, OMe), 3.79 (3H, s, OMe), 3.81 (3H, s, OMe), 6.64 (1H, d,  $J = 8.7$ , H-5'arom), 7.02 (1H, d,  $J = 8.7$ , H-6'arom), 7.1–7.2 (5H, m,  $H_{\text{arom}}$ ), 8.43 (1H, s, H-3);  $\delta_C$  55.6, 60.4, 60.7 (OMe), 107.0 ( $C_{\text{arom}}\text{H}$ ), 114.2 (C-4), 116.9 ( $C_{\text{arom}}\text{C}$ ), 125.1, 126.9, 127.1, and 128.3 ( $C_{\text{arom}}\text{H}$ ), 129.8 ( $C_{\text{arom}}\text{C}$ ), 140.0, 150.0, and 151.8 ( $C_{\text{arom}}\text{O}$ ), 155.3 (C-3), 162.3 (C-5);  $m/z$  311 ( $M^+$ , 4), 208 (42), 196 (45), 105 (100), 77 (33). Anal. Calcd for  $C_{18}H_{17}O_4N$ : C, 69.44; H, 5.50; N, 4.50. Found: C, 69.40; H, 5.43; N, 4.84.

**5-(3,4-Dimethoxyphenyl)-4-(3,4,5-trimethoxyphenyl)isoxazole (1f)** (95%): mp 52–54 °C (methanol);  $R_f$  ( $\text{CH}_2\text{Cl}_2$ –EtOAc, 9:1) 0.8;  $\nu_{\text{max}}$  1620 ( $\text{C}=\text{N}$ );  $\delta_{\text{H}}$  3.76 (6H, s, 2 × OMe), 3.79 (3H, s, OMe), 3.87 (3H, s, OMe), 3.89 (3H, s, OMe), 6.59 (2H, s, H-2'arom and H-6'arom), 6.85 (1H, d,  $J = 8.5$ , H-5'arom), 7.19 (1H, d,  $J = 2.0$ , H-2'arom), 7.26 (1H, dd,  $J = 8.5$ ,  $J = 2.0$ , H-6'arom), 8.30 (1H, s, H-3);  $\delta_C$  55.6, 55.7, 55.9, 60.8, and 61.2 (OMe), 107.2, 110.5, and 111.2 ( $C_{\text{arom}}\text{H}$ ), 114.8 (C-4), 117.1 ( $C_{\text{arom}}\text{C}$ ), 119.7 ( $C_{\text{arom}}\text{H}$ ), 122.7 ( $C_{\text{arom}}\text{C}$ ), 125.5 ( $C_{\text{arom}}\text{H}$ ), 142.3, 148.4, and 148.8 ( $C_{\text{arom}}\text{O}$ ), 150.2 (C-3), 152.2, 155.5 ( $C_{\text{arom}}\text{O}$ ), 161.9 (C-5). Anal. Calcd for  $C_{20}H_{21}O_6N$ : C, 64.68; H, 5.70; N, 3.77. Found: C, 64.62; H, 5.72; N, 4.01.

**4-(3,4,5-Trimethoxyphenyl)-5-(2,3,4-trimethoxyphenyl)isoxazole (1g)** (88%): colorless oil;  $R_f$  ( $\text{CH}_2\text{Cl}_2$ –EtOAc, 9:1) 0.9;  $\nu_{\text{max}}$  1610 ( $\text{C}=\text{N}$ );  $\delta_{\text{H}}$  3.69 (3H, s, OMe), 3.71 (6H, s, 2 × OMe), 3.82 (3H, s, OMe), 3.85 (3H, s, OMe), 3.89 (3H, s, OMe), 6.49 (2H, s, H-2'arom and H-6'arom), 6.72 (1H, d,  $J = 8.6$ , H-5'arom), 7.08 (1H, d,  $J = 8.6$ , H-6'arom), 8.47 (1H, s, H-3);  $\delta_C$  55.9, 56.1, 55.9, 60.8, 60.9, 61.4 (OMe), 104.6, 107.3 ( $C_{\text{arom}}\text{H}$ ), 114.7 (C-4), 117.1, 121.8 ( $C_{\text{arom}}\text{C}$ ), 125.6 ( $C_{\text{arom}}\text{H}$ ), 137.5, 142.4 ( $C_{\text{arom}}\text{O}$ ), 150 ( $C_{\text{arom}}\text{H}$ ), 142.3, 148.4, and 148.8 ( $C_{\text{arom}}\text{O}$ ), 150.2 (C-3), 152.3, 153.3, 155.6 ( $C_{\text{arom}}\text{O}$ ), 162.6 (C-5). Anal. Calcd for  $C_{21}H_{23}O_7N$ : C, 62.83; H, 5.77; N, 3.49. Found: C, 62.90; H, 5.72; N, 3.41.

**2,3-Diphenyl-3-oxopropanonitrile 4a. Typical Procedure.** NaH (106 mg, 4.4 mmol) was added in one portion to

a solution of isoxazole **1a** (114 mg, 0.52 mmol) in dry THF (25 mL) under argon at room temperature. The resulting slurry was refluxed for 10 h, cooled, quenched with saturated aqueous  $\text{NH}_4\text{Cl}$  (20 mL), and extracted with dichloromethane (3 × 20 mL). The combined organic layers were dried over anhydrous sodium sulfate and evaporated to give a yellow oil. Flash column chromatography (hexane/EtOAc 2:8) afforded a colorless oil, which was crystallized from hexane–diethyl ether, providing  $\beta$ -ketonitrile **4a** (80 mg, 70%) as a white solid: mp 91–92 °C (hexane–diethyl ether, 1:1) (lit.<sup>34</sup> mp 93–94 °C);  $R_f$  (hexane/EtOAc 2:8) 0.8;  $\nu_{\text{max}}$  2245 (CN), 1695 ( $\text{C}=\text{O}$ );  $\delta_{\text{H}}$  5.61 (3H, s, CHCN), 7.4–7.5 (8H, m,  $H_{\text{arom}}$ ), 7.9–8.0 (2H, m,  $H_{\text{arom}}$ );  $\delta_C$  46.6 (CHCN), 116.5 (CN), 128.2, 129.1, 129.2, 129.6, 130.3 ( $C_{\text{arom}}\text{H}$ ), 133.6, 134.4 ( $C_{\text{arom}}\text{C}$ ), 188.8 ( $\text{C}=\text{O}$ ). Anal. Calcd for  $C_{15}H_{11}ON$ : C, 81.43; H, 5.01; N, 6.33. Found: C, 81.20; H, 5.10; N, 6.40.

By use of the same procedure the following compounds were prepared:

**2,3-Bis(3,4-dimethoxyphenyl)-3-oxopropanonitrile (4b)** (91%): mp 103–105 °C (hexane/diethyl ether, 1:1);  $R_f$  (hexane/EtOAc 2:8) 0.8;  $\nu_{\text{max}}$  2240 (CN), 1685 ( $\text{C}=\text{O}$ );  $\delta_{\text{H}}$  3.84 (3H, s, OMe), 3.86 (3H, s, OMe), 3.89 (3H, s, OMe), 3.92 (3H, s, OMe), 5.51 (1H, s, CHCN), 6.83 (1H, d,  $J = 8.2$ , H-5''arom), 6.85 (1H, d,  $J = 8.4$ , H-5''arom), 6.91 (1H, d,  $J = 2.2$ , H-2''arom), 6.97 (1H, dd,  $J = 8.2$ , 2.2, H-6''arom), 7.49 (1H, d,  $J = 2.1$ , H-2''arom), 7.57 (1H, dd,  $J = 8.2$ ,  $J = 2.1$ , H-6''arom);  $\delta_C$  45.9 (CHCN), 55.9, 56.0, 56.2 (OMe), 110.1, 111.2, 111.6 ( $C_{\text{arom}}\text{H}$ ), 116.9 (CN), 120.7 ( $C_{\text{arom}}\text{H}$ ), 123.0 ( $C_{\text{arom}}\text{C}$ ), 124.1 ( $C_{\text{arom}}\text{H}$ ), 126.5 ( $C_{\text{arom}}\text{C}$ ), 149.3, 149.6, 149.8, 154.3 ( $C_{\text{arom}}\text{O}$ ), 187.5 ( $\text{C}=\text{O}$ ). Anal.  $C_{19}H_{19}O_5N$ : C, 66.85; H, 5.61; N, 4.10. Found: C, 66.81; H, 5.50; N, 4.19.

**3-(3,4-Dimethoxyphenyl)-2-phenyl-3-oxopropanonitrile (4c)** (91%): mp 113–114 °C (hexane/diethyl ether, 1:1);  $R_f$  (hexane/EtOAc 2:8) 0.5;  $\nu_{\text{max}}$  2245 (CN), 1685 ( $\text{C}=\text{O}$ );  $\delta_{\text{H}}$  3.77 (3H, s, 2 × OMe), 3.81 (3H, s, OMe), 5.62 (1H, s, CHCN), 6.77 (1H, d,  $J = 8.4$ , H-5'arom), 7.2–7.3 (5H, m,  $H_{\text{arom}}$ ), 7.40 (1H, d,  $J = 1.8$ , H-2'arom), 7.52 (1H, dd,  $J = 8.4$ , 1.8, H-6'arom);  $\delta_C$  46.0 (CHCN), 55.7, 55.9 (OMe), 110.0, 110.9, ( $C_{\text{arom}}\text{H}$ ), 116.9 (CN), 126.6, ( $C_{\text{arom}}\text{C}$ ), 127.9, 128.8, 129.3 ( $C_{\text{arom}}\text{H}$ ), 130.8 ( $C_{\text{arom}}\text{C}$ ), 149.1, 154.2 ( $C_{\text{arom}}\text{O}$ ), 187.4 ( $\text{C}=\text{O}$ ). Anal. Calcd for  $C_{17}H_{15}O_3N$ : C, 72.58; H, 5.37; N, 4.98. Found: C, 72.66; H, 5.30; N, 5.01.

**2-(3,4-Dimethoxyphenyl)-3-(2,3,4-trimethoxyphenyl)-3-oxopropanonitrile (4d)** (88%): mp 114–115 °C (hexane/diethyl ether, 1:1);  $R_f$  (hexane/EtOAc 2:8) 0.3;  $\nu_{\text{max}}$  2245 (CN), 1695 ( $\text{C}=\text{O}$ );  $\delta_{\text{H}}$  3.69 (3H, s, OMe), 3.70 (3H, s, OMe), 3.71 (3H, s, OMe), 3.74 (3H, s, OMe), 3.86 (3H, s, OMe), 5.74 (1H, s, CHCN), 6.54 (1H, d,  $J = 8.9$ , H-5'arom), 6.66 (1H, d,  $J = 8.1$ , H-5''arom), 6.74 (1H, d,  $J = 2.0$ , H-2''arom), 6.77 (1H, dd,  $J = 8.1$ , 2.0, H-6''arom), 7.28 (1H, d,  $J = 8.9$ , H-6'arom);  $\delta_C$  48.5 (CHCN), 55.6, 55.7, 55.9, 60.6, 61.7 (OMe), 107.2, 110.9, 111.1 ( $C_{\text{arom}}\text{H}$ ), 117.2 (CN), 120.8 ( $C_{\text{arom}}\text{C}$ ), 121.8 ( $C_{\text{arom}}\text{H}$ ), 122.9 ( $C_{\text{arom}}\text{C}$ ), 126.7 ( $C_{\text{arom}}\text{H}$ ), 141.2, 149.1, 153.1, 158.3 ( $C_{\text{arom}}\text{O}$ ), 189.5 ( $\text{C}=\text{O}$ ). Anal. Calcd for  $C_{20}H_{21}O_6N$ : C, 64.68; H, 5.70; N, 3.77. Found: C, 64.40; H, 5.61; N, 3.89.

**N-Methyl-4,5-diphenylisoxazolium Tetrafluoroborate (6a). Typical Procedure.** Isoxazole **1a** (500 mg, 2.2 mmol) was dissolved in freshly distilled dichloromethane and added dropwise into a stirred suspension of trimethyloxonium tetrafluoroborate (500 mg, 3.3 mmol) in dichloromethane under argon at room temperature. After being stirred overnight, the reaction mixture was evaporated *in vacuo* and the resulting oil was treated with diethyl ether (2 mL) to give a yellow powder, which was filtered and crystallized in ethyl acetate, affording **6a** (600 mg, 85%) as white needles:<sup>35</sup> mp 139–141 °C (ethyl acetate);  $R_f$  ( $\text{CH}_2\text{Cl}_2$ ) 0.5;  $\nu_{\text{max}}$  1645 ( $\text{C}=\text{N}^+$ );  $\delta_{\text{H}}$  (DMSO-*d*<sub>6</sub>) 4.63 (3H, s, NMe), 7.4–7.7 (10H, m,  $H_{\text{arom}}$ ), 9.76 (1H, s, H-3);  $\delta_C$  (DMSO-*d*<sub>6</sub>) 45.1 (NMe), 125.3 ( $C_4$ ), 128.7, 130.9, 127.7, and 134.7 ( $C_{\text{arom}}\text{H}$ ), 154.1 (C-3), 172.8 (C-5). Anal. Calcd for  $C_{16}H_{14}ONBF_4$ : C, 59.48; H, 4.37; N, 4.33. Found: C, 59.40; H, 4.41; N, 4.39.

(34) Kascheres, A.; Marchi, D. *J. Org. Chem.* **1975**, *40*, 2985–2987.

(35) Similarly to some other isoxazolium salts of the literature (see ref 23), **6a** turned yellow and decomposed in 1–2 months at room temperature.

By use of the same procedure the following compounds were prepared:

**N-Methyl-4,5-bis(3,4-dimethoxyphenyl)isoxazolium tetrafluoroborate (6b)** (94%): white powder; mp 161–163 °C (ethyl acetate);  $R_f$  (CH<sub>2</sub>Cl<sub>2</sub>) 0.5;  $\nu_{\max}$  1650 (C=N<sup>+</sup>);  $\delta_H$  (DMSO-*d*<sub>6</sub>) 3.71 (3H, s, OMe), 3.77 (3H, s, OMe), 3.87 (3H, s, OMe), 3.91 (3H, s, OMe), 4.60 (3H, s, NMe), 7.10 (1H, d,  $J = 8.3$ , H-5'<sub>arom</sub> and H-6''<sub>arom</sub>), 7.33 (1H, d,  $J = 2.1$ , H-2'<sub>arom</sub>), 7.43 (1H, dd,  $J = 8.3$ ,  $J = 2.1$ , H-6'<sub>arom</sub>), 9.61 (1H, s, H-3);  $\delta_C$  40.9 (NMe), 55.6, 55.9 (OMe), 110.3, 112.2 (C<sub>arom</sub>H), 115.2 (C-4), 117.3, 117.9 (C<sub>arom</sub>H), 121.5 (C<sub>arom</sub>C), 149.0, 149.2, and 150.0 (C<sub>arom</sub>O), 152.9 (C-3), 165.6 (C-5). Anal. Calcd for C<sub>20</sub>H<sub>22</sub>O<sub>5</sub>NBF<sub>4</sub>: C, 54.20; H, 5.00; N, 3.16. Found: C, 54.29; H, 5.08; N, 3.20.

**N-Methyl-5-(3,4-dimethoxyphenyl)-4-phenylisoxazolium tetrafluoroborate (6c)** (96%): white powder; mp 158–160 °C (ethyl acetate);  $R_f$  (CH<sub>2</sub>Cl<sub>2</sub>) 0.2;  $\nu_{\max}$  1650 (C=N<sup>+</sup>);  $\delta_H$  (DMSO-*d*<sub>6</sub>) 3.66 (3H, s, OMe), 3.90 (3H, s, OMe), 4.60 (3H, s, NMe), 7.12 (1H, d,  $J = 8.6$ , H-5'<sub>arom</sub>), 7.25 (1H, d,  $J = 2.1$ , H-2'<sub>arom</sub>), 7.38 (1H, dd,  $J = 8.6$ , 2.1, H-6'<sub>arom</sub>), 7.59 (5H, m, H<sub>arom</sub>), 7.04 (1H, s, H-2'<sub>arom</sub>), 7.15 (1H, dd,  $J = 8.4$ , 1.9, H-6''<sub>arom</sub>), 7.21 (1H, d,  $J = 1.9$ , H-2''<sub>arom</sub>), 9.60 (1H, s, H-3);  $\delta_C$  (DMSO-*d*<sub>6</sub>) 41.5 (NMe), 56.0, 56.3 (OMe), 111.4, 112.8 (C<sub>arom</sub>H), 116.2 (C-4), 119.3, 123.4 (C<sub>arom</sub>H), 127.1, 130.3 (C<sub>arom</sub>C), 150.5 (C<sub>arom</sub>O), 154.6 (C-3), 186.3 (C-5). Anal. Calcd for C<sub>18</sub>H<sub>18</sub>O<sub>3</sub>NBF<sub>4</sub>: C, 56.43; H, 4.73; N, 3.65. Found: C, 56.39; H, 4.70; N, 3.70.

**N-Methyl-4-(3,4-dimethoxyphenyl)-5-(2,3,4-trimethoxyphenyl)isoxazolium tetrafluoroborate (6d)** (99%): white powder; mp 152–153 °C (ethyl acetate);  $R_f$  (CH<sub>2</sub>Cl<sub>2</sub>) 0.6;  $\nu_{\max}$  1670 (C=N<sup>+</sup>);  $\delta_H$  (DMSO-*d*<sub>6</sub>) 3.34 (3H, s, OMe), 3.60 (3H, s, OMe), 3.67 (3H, s, OMe), 3.77 (3H, s, OMe), 3.90 (3H, s, OMe), 4.45 (3H, s, NMe), 6.92 (1H, dd,  $J = 8.4$ , 1.9, H-6''<sub>arom</sub>), 7.00 (1H, d,  $J = 1.9$ , H-2''<sub>arom</sub>), 7.05 (1H, d,  $J = 8.4$ , H-5''<sub>arom</sub>), 7.06 (1H, d,  $J = 8.9$ , H-5'<sub>arom</sub>), 7.31 (1H, d,  $J = 8.9$ , H-6'<sub>arom</sub>), 10.09 (1H, s, H-3);  $\delta_C$  (DMSO-*d*<sub>6</sub>) 41.2 (NMe), 55.5, 56.3, 60.7, and 61.3 (OMe), 109.2, 111.0 (C<sub>arom</sub>H), 118.1 (C-4), 120.5 (C<sub>arom</sub>H), 126.2, 141.6 (C<sub>arom</sub>C), 148.9, 149.8, and 151.9 (C<sub>arom</sub>O), 157.7 (C-3), 164.1 (C-5). Anal. Calcd for C<sub>21</sub>H<sub>24</sub>O<sub>6</sub>NBF<sub>4</sub>: C, 53.30; H, 5.11; N, 2.96. Found: C, 53.24; H, 5.12; N, 2.99.

**N-Methyl-4,5-diphenyl-4-isoxazoline (5a). Typical Procedure.** A stirred suspension of NaBH<sub>4</sub> (110 mg, 2.9 mmol) in dry ethanol (15 mL) was added *via cannula* to a stirred suspension of isoxazolium salt **6a** (108 mg, 0.33 mmol) in the same solvent (5 mL) under argon at room temperature. After being stirred for 6 h the reaction was quenched by adding saturated aqueous NH<sub>4</sub>Cl solution (10 mL) and extracted with dichloromethane (5 × 20 mL). The aqueous layer was acidified to pH 8 with diluted hydrochloric acid and extracted again with dichloromethane (5 × 20 mL). The combined organic layers were washed with water, dried over anhydrous sodium sulfate, and evaporated under reduced pressure to give a brown oil that was purified by flash chromatography using CH<sub>2</sub>Cl<sub>2</sub> as eluent. Isoxazoline **5a** was obtained (66 mg, 84%) as a colorless oil;  $R_f$  (CH<sub>2</sub>Cl<sub>2</sub>) 0.5;  $\nu_{\max}$  1661 (C=C);  $\delta_H$  2.92 (3H, s, NMe), 3.99 (1H, m, CH<sub>a</sub>H<sub>b</sub>N), 4.75 (1H, m, CH<sub>a</sub>H<sub>b</sub>N), 7.1–7.3 (8H, m, H<sub>arom</sub>), 7.51 (2H, m, H<sub>arom</sub>);  $\delta_C$  47.5 (NMe), 65.7 (CH<sub>2</sub>), 105.8 (C<sub>4</sub>), 126.4, 126.9, and 128.3 (C<sub>arom</sub>H), 128.4 (C<sub>arom</sub>C), 128.4, 129.2 (C<sub>arom</sub>H), 133.6 (C<sub>arom</sub>C), 147.3 (C-5). Anal. Calcd for C<sub>16</sub>H<sub>15</sub>ON: C, 80.98; H, 6.37; N, 5.90. Found: C, 81.06; H, 6.36; N, 5.92.

By use of the same procedure the following compounds were prepared:

**N-Methyl-4,5-bis(3,4-dimethoxyphenyl)-4-isoxazoline (5b)** (64%): white powder; mp 109–111 °C (methanol);  $R_f$  (CH<sub>2</sub>Cl<sub>2</sub>) 0.3;  $\nu_{\max}$  1655 (C=C);  $\delta_H$  2.82 (3H, s, NMe), 3.59 (3H, s, OMe), 3.67 (3H, s, OMe), 3.77 (3H, s, OMe), 3.78 (3H, s, OMe), 3.96 (1H, m, CH<sub>a</sub>H<sub>b</sub>N), 4.59 (1H, m, CH<sub>a</sub>H<sub>b</sub>N), 6.6–6.7 (4H, m, H<sub>arom</sub>), 7.51 (2H, m, H<sub>arom</sub>);  $\delta_C$  47.2 (NMe), 55.3, 55.5, 55.6 (OMe), 65.5 (C<sub>3</sub>), 105.8 (C<sub>4</sub>), 109.9, 110.5, 110.8, 119.1 and 120.9 (C<sub>arom</sub>H), 121.7, 126.1 (C<sub>arom</sub>C), 145.7, 147.4, 148.2,

and 148.3 (C<sub>arom</sub>O), 149.3 (C-5). Anal. Calcd for C<sub>20</sub>H<sub>23</sub>O<sub>5</sub>N: C, 67.21; H, 6.49; N, 3.92. Found: C, 67.29; H, 6.44; N, 3.92.

**N-Methyl-5-(3,4-dimethoxyphenyl)-4-phenyl-4-isoxazoline (5c)** (90%): colorless oil;  $R_f$  (CH<sub>2</sub>Cl<sub>2</sub>) 0.4;  $\nu_{\max}$  1660 (C=C);  $\delta_H$  2.85 (3H, s, NMe), 3.65 (3H, s, OMe), 3.80 (3H, s, OMe), 3.99 (1H, m, CH<sub>a</sub>H<sub>b</sub>N), 4.65 (1H, m, CH<sub>a</sub>H<sub>b</sub>N), 6.74 (1H, d,  $J = 8.3$ , H-5'<sub>arom</sub>), 6.95 (1H, d,  $J = 1.9$ , H-2'<sub>arom</sub>), 7.05 (1H, dd,  $J = 8.3$ , 1.9, H-6'<sub>arom</sub>), 7.1–7.2 (5H, m, H<sub>arom</sub>);  $\delta_C$  47.1 (NMe), 55.3, 55.4 (OMe), 65.5 (C<sub>3</sub>), 104.4 (C<sub>4</sub>), 110.5, 110.7 (C<sub>arom</sub>H), 120.8 (C<sub>arom</sub>C), 121.4, 125.9, 126.6, and 127.9 (C<sub>arom</sub>H), 133.5 (C<sub>arom</sub>C), 146.7, 148.2 (C<sub>arom</sub>O), 149.4 (C-5). Anal. Calcd for C<sub>18</sub>H<sub>19</sub>O<sub>3</sub>N: C, 72.71; H, 6.44; N, 4.71. Found: C, 72.79; H, 6.36; N, 4.80.

**N-Methyl-4-(3,4-dimethoxyphenyl)-5-(2,3,4-trimethoxyphenyl)-4-isoxazoline (5d)** (93%): colorless oil;  $R_f$  (CH<sub>2</sub>Cl<sub>2</sub>) 0.3;  $\nu_{\max}$  1645 (C=C);  $\delta_H$  2.84 (3H, s, NMe), 3.47 (3H, s, OMe), 3.72 (3H, s, OMe), 3.76 (3H, s, OMe), 3.78 (3H, s, OMe), 3.79 (3H, s, OMe), 3.82 (1H, m, CH<sub>a</sub>H<sub>b</sub>N), 4.64 (1H, m, CH<sub>a</sub>H<sub>b</sub>N), 6.50 (1H, d,  $J = 8.3$ , H-5'<sub>arom</sub>), 6.52 (1H, s, H-2''<sub>arom</sub>), 6.61 (1H, d,  $J = 5.7$ , H-6''<sub>arom</sub>), 6.65 (1H, d,  $J = 5.7$ , H-5''<sub>arom</sub>), 6.93 (1H, d,  $J = 8.3$ , H-6'<sub>arom</sub>);  $\delta_C$  47.4 (NMe), 54.8, 55.4, 55.7, 60.6, and 61.1 (OMe), 63.8 (C<sub>3</sub>), 106.5, 107.3, 108.5, and 110.7 (C<sub>arom</sub>H), 116.5 (C<sub>4</sub>), 117.7 (C<sub>arom</sub>C), 125.7 (C<sub>arom</sub>H), 125.8 (C<sub>arom</sub>C), 142.4, 145.1, 146.9, 148.1, 152.6 (C<sub>arom</sub>O), 154.7 (C-5). Anal. Calcd for C<sub>21</sub>H<sub>25</sub>O<sub>6</sub>N: C, 65.10; H, 6.50; N, 3.61. Found: C, 65.04; H, 6.58; N, 3.60.

**3-(N-Methylamino)-1,2-diphenylpropanol (7a). Typical Procedure.** A mixture of 4-isoxazoline **5a** (95 mg, 0.4 mmol), prerduced Adams catalyst (from 10 mg of platinum oxide), and HCl (0.5 mL of a 3 mol L<sup>-1</sup> solution in water) in dry methanol (20 mL) was hydrogenated for 7 h (P<sub>H<sub>2</sub></sub> = 2.5 atm). The mixture was filtered, and the filtrate was evaporated to give a brown oil that was dissolved in diethyl ether and treated with 3 drops of concentrated HCl solution. The resulting solid hydrochloride was filtered, washed with dichloromethane, and redissolved in a mixture of methanol–water (9:1). This solution was basified to pH 9 with ammonium hydroxide solution to produce the free amino alcohol and extracted with dichloromethane (4 × 15 mL). The combined organic extracts were dried over anhydrous sodium sulfate and evaporated in vacuo to afford 1,3-amino alcohol **7a** (45 mg, 47%): colorless oil;  $R_f$  (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 9.5:0.5) 0.7;  $\nu_{\max}$  3500 (OH), 3300 (NH);  $\delta_H$  2.33 (3H, s, NMe), 2.51 (1H, bs, NH or OH), 2.83 (2H, d,  $J = 6.5$ , CH<sub>2</sub>N), 3.29 (1H, q,  $J = 6.5$ , CHCH<sub>2</sub>N), 4.93 (1H, d,  $J = 6.5$ , CHOH), 7.1–7.2 (10H, m, H<sub>arom</sub>);  $\delta_C$  36.1 (NMe), 51.5 (CHCH<sub>2</sub>N), 53.0 (CH<sub>2</sub>N), 77.9 (CHOH), 126.6, 126.8, 127.1, 127.8, 128.3, 128.4, and 129.4 (C<sub>arom</sub>H), 139.7, 141.9 (C<sub>arom</sub>C). Anal. Calcd for C<sub>16</sub>H<sub>19</sub>ON: C, 79.63; H, 7.93; N, 5.80. Found: C, 79.66; H, 7.98; N, 5.72.

By use of the same procedure 3-(N-methylamino)-1,2-bis(3,4-dimethoxyphenyl)propanol (**7b**) was prepared (55%): pale yellow powder; mp 118–119 °C (dichloromethane);  $R_f$  (CH<sub>2</sub>Cl<sub>2</sub>/MeOH 9.5:0.5) 0.4;  $\nu_{\max}$  3510 (OH), 3310 (NH);  $\delta_H$  2.33 (3H, s, NMe), 2.61 (1H, bs, NH or OH), 2.78 (2H, d,  $J = 6.7$ , CH<sub>2</sub>N), 3.22 (1H, q,  $J = 6.7$ , CHCH<sub>2</sub>N), 3.74 (3H, s, OMe), 3.78 (3H, s, OMe), 3.84 (6H, s, OMe), 4.85 (1H, d,  $J = 6.7$ , CHOH), 6.6–6.8 (6H, m, H<sub>arom</sub>);  $\delta_C$  36.3 (NMe), 51.4 (CHCH<sub>2</sub>N), 54.3 (CH<sub>2</sub>N), 55.6, 55.8 (OMe), 78.1 (CHOH), 109.9, 110.3, 111.1, 111.9, 119.1, and 120.6 (C<sub>arom</sub>H), 132.2, 134.4 (C<sub>arom</sub>C), 147.8, 148.2, 148.3, and 148.7 (C<sub>arom</sub>O). Anal. Calcd for C<sub>20</sub>H<sub>27</sub>O<sub>5</sub>N: C, 66.46; H, 7.53; N, 3.87. Found: C, 66.50; H, 7.55; N, 3.91.

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